

STRUCTURE OF AN ANTITUMOR
ANTIBIOTIC, SPERGUALIN

Sir:

As reported in the preceding paper¹⁾, in the study of antitumor antibiotics, an antibiotic which exhibited a strong inhibition against experimental mouse tumors has been discovered in bacterial culture filtrates and named spergualin. In this paper, the structural elucidation of this antibiotic is described.

Spergualin (**1**) trihydrochloride was obtained from culture filtrates of the bacterial strain BMG162-aF2 by the processes described in the preceding paper. It shows no definite melting point and $[\alpha]_D^{24} -11^\circ$ (*c* 1, H₂O). Anal. Calcd. for C₁₇H₃₇N₇O₄·3HCl·½H₂O: C 39.12, H 7.92, N 18.79, Cl 20.38. Found: C 39.00, H 8.02, N 17.40, Cl 19.65. UV (H₂O); end absorption, IR (KBr); 3400, 1660, 1540, 1470, 1170, 1115, 1090 and 1020 cm⁻¹. ¹H NMR (D₂O); δ 1.8~2.3 (CH₂×5), 2.57 (2-CH₂), 2.95 (14-CH₂, *J*=1, 3.5, 8 Hz), 3.5~3.8 (NCH₂×5), 4.55 (15-CH), 5.98 (11-CH), ¹⁵N NMR (D₂O, ppm relative to exter-

nal NO₃⁻); -240 (CONH), -257 (CONH), -290 (NH), -304 (NH=C-NH₂), -330 (NH), -343 (NH₂). The chemical shifts of the ¹³C NMR spectra are shown in Table 1. The antibiotic (**1**) gives positive ninhydrin, RYDON-SMITH and SAKAGUCHI reactions. It is easily soluble in water and methanol, but slightly or not in ethanol, ethyl acetate, acetone, cyclohexane and other organic solvents. The thin-layer chromatography on a cellulose plate (Avicel) developed with 1-butanol-pyridine-acetic acid-water (6:4:1:3) and with 1-butanol-ethanol-water (4:1:2) showed R_f 0.14 and 0.20, respectively. By high-voltage paper electrophoresis with 3,500 V for 15 minutes in formic acid-acetic acid-water (1:3:36), **1** moved to the cathode with R_m (relative mobility to alanine) 1.6.

The crystalline tripicrate of **1** was obtained from the aqueous ethanol solution, mp 62~78°C (decomp.). Anal. Calcd. for C₁₇H₃₇N₇O₄·3C₆H₃N₃O₇: C 38.54, H 4.25, N 20.54, picric acid 63.01. Found: C 38.76, H 4.71, N 18.31, picric acid (UV) 60.0.

Acid hydrolysis of **1** (trihydrochloride, 74.8mg)

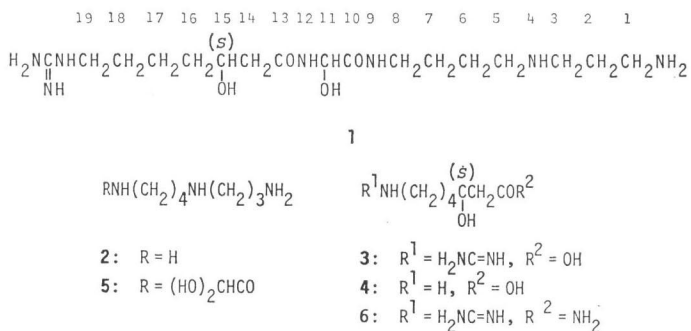


Table 1. ¹³C Chemical shifts (ppm) of spergualin (**1**) and compounds **3**, **5** and **6**.

Carbon	1·3HCl	3·HCl	5·2HCl	6·HCl	Carbon	1·3HCl	3·HCl	5·2HCl	6·HCl
13	174.0 s	178.0		173.3	1, 8	{ 38.1 t		37.8	
10	170.6 s		171.7			{ 36.2 t		36.3	
C=N	156.2 s	157.5		156.5	16	35.0 t	36.2		36.7
11	71.4 d		86.3		18	27.2 t	28.5		28.3
15	67.9 d	69.0		67.0	2	25.0 t		25.0	
3, 5	{ 46.2 t		46.9		6, 7	{ 23.2 t		23.3	
	{ 44.0 t		44.0			{ 22.4 t		22.4	
19	42.8 t	42.2		43.0	17	21.4 t	22.7		22.1
14	40.6 t	41.8		41.7					

The ¹³C FT NMR spectra were taken with a Varian XL-100 spectrometer in D₂O. Multiplicities (s, d, and t) were shown by off-resonance experiment.

with 1 N HCl in a sealed tube at 105°C for 3 hours followed by column chromatography on Sephadex LH-20 developed with methanol gave a polyamine **2** as the hydrochloride (40.9 mg), which was identical with authentic spermidine trihydrochloride in all respects, and a hydrochloride of a new SAKAGUCHI-positive compound **3** (28.7 mg), $[\alpha]_D^{25} + 2^\circ$ (*c* 2.1, 1 N HCl), $C_8H_{17}N_3O_3 \cdot HCl$, FD-MS; *m/z* 204 (MH⁺), ¹H NMR (D₂O); δ 1.8~2.3 (4-, 5-, 6-CH₂), 2.95 (2-CH₂, *J*=2, 7 Hz), 3.66 (7-CH₂, *t*, *J*=7 Hz), 4.5 (3-CH). The structure of **3** was determined to be 7-guandinio-3-hydroxyheptanoic acid by spectral data of ¹H and ¹³C NMR (Table 1). Hydrolysis of **3** with Ba(OH)₂-saturated aqueous solution gave a deguanidino compound **4** as the hydrochloride, $[\alpha]_D^{25} + 3^\circ$ (*c* 0.95, 1 N HCl). The compound **4** hydrochloride was identical with the hydrochloride of (S)-7-amino-3-hydroxyheptanoic acid which was obtained by chemical synthesis starting from L-lysine²⁾. Therefore, the C-3 of **3** has *S*-configuration.

Mild acid hydrolysis of **1** (trihydrochloride, 49.0 mg) with 1 M acetic acid for 1.5 hours under refluxing followed by column chromatography on Sephadex LH-20 developed with methanol gave a ninhydrin-positive compound **5** (hydrochloride, 31.3 mg) $C_9H_{21}N_3O_3 \cdot 2HCl$, ¹H NMR (D₂O); δ 2.1 (6-, 7-CH₂), 2.4~2.7 (2-CH₂), 3.5~3.8 (1-, 3-, 5-, 8-CH₂), 5.77 (11-CH, *s*) and a SAKAGUCHI-positive compound **6** (hydrochloride, 19.7 mg), $[\alpha]_D^{25} - 3^\circ$ (*c* 1, H₂O), $C_8H_{18}N_4O_2 \cdot HCl$, FD-MS; *m/z* 203 (MH⁺), ¹H NMR (D₂O); δ 1.8~2.3 (4-, 5-, 6-CH₂), 2.95 (2-CH₂, *J*=2, 6 Hz), 3.69 (7-CH₂, *t*, *J*=7 Hz), 4.5 (3-CH), which was shown to be an amide of **3**. Treatment of **5** (hydrochloride, 24.7 mg) with 2,4-dinitrophenylhydrazine (25 mg) in 1 N HCl under refluxing for 1.5 hours followed by extraction with ethyl acetate gave yellow needles which was identified to be glyoxylic acid 2,4-dinitrophenylhydrazone (2.4 mg) by comparison with an authentic sample³⁾ (IR and TLC). From the aqueous layer of the ethyl acetate extraction, 2,4-dinitrophenylhydrazone of **5** (28.7 mg) was obtained as its dihydrochloride, $C_{15}H_{23}N_7O_5 \cdot 2HCl \cdot H_2O$, FD-MS; *m/z* 382 (MH⁺). By deamination of the hydrazone with NaNO₂ in acetic acid followed by acid hydrolysis with 6 N HCl, *N*-(3-hydroxypropyl)-1,4-butanediamine hydrochloride was obtained and identified with a synthetic sample⁴⁾ by their ¹H NMR spectra. Therefore, the struc-

ture of **5** was determined to be 11-amino-1,1-dihydroxy-3,8-diazaundecan-2-one (a hydrate of glyoxylylspermidine).

From the foregoing results and ¹³C NMR data (Table 1), the structure of spergualin (**1**) can be proposed to be (–)-(15*S*)-1-amino-19-guanidino-11,15-dihydroxy-4,9,12-triazanonadecane-10,13-dione. The total synthesis of **1** including the synthesis of **4** will be reported in the next paper²⁾.

Among known antibiotics, laterosporamine⁵⁾ is similar to spergualin in its optical rotation, molecular formula ($C_{17}H_{35}N_7O_4$), color reactions and weak antibacterial activities. However, no antitumor activity has been reported for laterosporamine. Although it was reported that the hydrolysis of laterosporamine gave spermidine and a SAKAGUCHI-positive substance of $C_6H_{13}N_3O$, the hydrolysis of spergualin did not give such a C₆-compound. Moreover, the description on laterosporamine suggests that laterosporamine hydrochloride has a stronger activity against *Staphylococcus aureus* FDA 209P than spergualin trihydrochloride¹⁾ and also is more soluble in ethanol.

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HAMAO UMEZAWA
SHINICHI KONDO
HIRONOBU IINUMA
SETSUKO KUNIMOTO
YOKO IKEDA
HIROYUKI IWASAWA
DAISHIRO IKEDA
TOMIO TAKEUCHI

Institute of Microbial Chemistry
14-23 Kamiosaki 3-Chome,
Shinagawa-ku, Tokyo 141, Japan

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